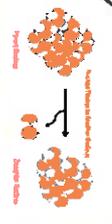


Radius of an atom  
 $1 \times 10^{-10}m$



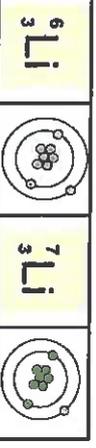
Electrons gained  
Negative ion

Electrons lost  
Positive ion



Atom	Same number of protons and electrons	
Ion	Unequal number of electrons to protons	
Mass number	Number of protons and neutrons	
Atomic number	Number of protons	

Particle	Charge	Size	Found
Neutron	None	1	In the nucleus
Proton	+	1	
Electron	-	Tiny	Orbits the nucleus



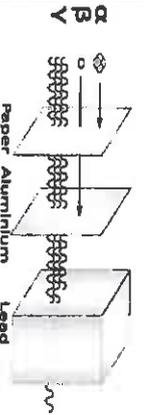
Different forms of an element with the same number of protons but different number of neutrons

Discovery of the nucleus

Democritus	Suggested idea of atoms as small spheres that cannot be cut.
J J Thomson (1897)	Discovered electrons—emitted from surface of hot metal. Showed electrons are negatively charged and that they are much less massive than atoms.
Thomson (1904)	Proposed 'plum pudding' model – atoms are a ball of positive charge with negative electrons embedded in it.
Gelger and Marsden (1909)	Directed beam of alpha particles ( $He^{2+}$ ) at a thin sheet of gold foil. Found some travelled through, some were deflected, some bounced back.
Rutherford (1911)	Used above evidence to suggest alpha particles deflected due to electrostatic interaction between the very small charged nucleus. Nucleus was massive. Proposed mass and positive charge contained in nucleus while electrons found outside the nucleus which cancel the positive charge exactly.
Bohr (1913)	Suggested modern model of atom—electrons in circular orbits around nucleus, electrons can change orbits by emitting or absorbing electromagnetic radiation. His research led to the idea of some particles within the nucleus having positive charge; these were named protons.
Chadwick (1932)	Discovered neutrons in nucleus – enabling other scientists to account for mass of atom.

Radioactive decay	Unstable atoms randomly emit radiation to become stable
Detecting	Use Geiger Muller tube
Unit	Becquerel
Ionisation	All radiation ionises

Decay	Range in air	Ionising power	Penetration power
Alpha	Few cm	Very strong	Stopped by paper
Beta	Few m	Medium	Stopped by Aluminium
Gamma	Great distances	Weak	Stopped by thick lead



Atoms and Isotopes  
Atoms and Nuclear Radiation

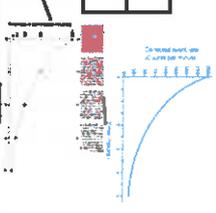
AQA ATOMIC STRUCTURE

PHYSICS ONLY: Hazards and uses of Radioactive emissions and of background radiation

Decay	Emitted from nucleus	Changes in mass number and atomic number
Alpha ( $\alpha$ )	Helium nuclei ( ${}^4_2He$ )	-4      -2
Beta ( $\beta$ )	Electron ( ${}^0_{-1}e$ )	0      +1
Gamma ( $\gamma$ )	Electromagnetic wave	0      0
Neutron	Neutron	-1      0

Contamination: Unwanted presence of radioactive atoms  
Irradiation: Person is in exposed to radioactive source

Half life: The time taken to lose half of its initial radioactivity



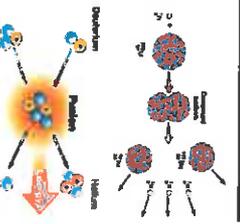
Sievert: Unit measuring dose of radiation  
Background: Constant low level environmental radiation, e.g. from nuclear testing, nuclear power, waste

Uses	Different isotopes have different half lives	Short half-lives used in high doses, long half lives used in low doses.
Tracers	Used within body	Isotope with short half life injected, allowed to circulate and collect in damaged areas. PET scanner used to detect emitting radiation. Must be beta or gamma as alpha does not penetrate the body.
Radiation therapy	Used to treat illnesses e.g. cancer	Cancer cells killed by gamma rays. High dose used to kill cells. Damage to healthy cells prevented by focussed gamma ray gun.

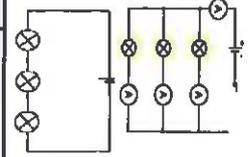
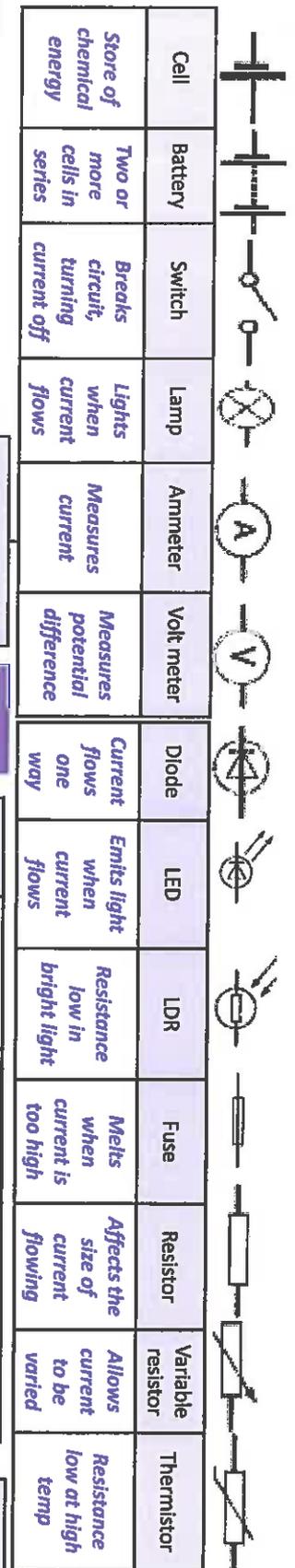
Nuclear fission and fusion

PHYSICS ONLY: Nuclear energy

Nuclear fission	Nuclear fusion
One large unstable nucleus splits to make two smaller nuclei	Two small nuclei join to make one larger nucleus
Neutron hits U-235 nucleus, nucleus absorbs neutron, splits emitting two or three neutrons and two smaller nuclei. Process also releases energy.	Difficult to do on Earth – huge amounts of pressure and temperature needed.
Control rods: Made of Boron. Controls the rate of reaction. Boron absorbs excess neutrons.	Process repeats, chain reaction formed
Fuel rods: Made of U-238, 'enriched' with U-235 (3%). Long and thin to allow neutrons to escape, hitting nuclei.	Used in nuclear power stations
Concrete: Neutrons hazardous to humans – thick concrete shield protects workers.	Occurs in stars



Electrons carry current.  
Electrons are free to move in metal.



**Circuit symbols**

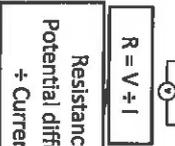
Current	Flow of electrical charge	Ampere (A)
Potential difference (p.d.)	How much electrical work is done by a cell	Volts (V)
Charge	Amount of electricity travelling in a circuit	Coulombs (C)

Charge = Current X time  
 $Q = I \times t$

Controlling current

Ammeter: Set up in series with components  
Voltmeter: Set up parallel to components

Resistance ( $\Omega$ ): A measurement of how much current flow is reduced  
The higher the resistance, the more difficult it is for current to flow.  
Increasing resistance, reduces current.  
Increasing voltage, increases current.



**Current, potential difference and resistance**

**Series and parallel circuits**

Series circuit	Current is the same in all components.	Total p.d. from battery is shared between all the components.	Total resistance is the sum of each component's resistance.
Parallel circuit	Total current is the sum of each component's current.	p.d. across all components is the same.	Total resistance is less than the resistance value of the smallest individual resistor.

**Series**: A circuit with one loop  
**Parallel**: A circuit with two or more loops  
If cells are joined in series, add up individual cell values

**Electricity**  
AQA  
Domestic uses and safety

**Energy transfers**  
Work is done when charge flows.  
National Grid: Distributes electricity generated in power stations around UK

**Static electricity**  
When two insulating material are rubbed together, electrons move from one material to the other.  
PHYSICS only

Ohmic conduct or Filament lamp	At a constant temperature, current is directly proportional to the p.d. across the resistor. As current increases, the resistance increases. The temperature increases as current flows.
Diode	Current flows when p.d. flows forward. Very high resistance in reverse

**Current - Potential difference graphs**

Thermistor	Resistance varies with temperature Resistance decreases as temperature increases.
LDR	Resistance varies with light intensity Resistance decreases as light increases.

'Earthing' a safety device; Earth wire joins the metal case.  
Mains supply: Frequency 50Hz, 230V  
Live - Brown: Carries p.d. from mains supply.  
Neutral - Blue: Completes the circuit.  
Earth - Green and Yellow stripes: Only carries current if there is a fault.

Generator: p.d. remains in one direction, current flows the same direction  
Cell or battery: p.d. = 0V

**Shocks**  
Walking on carpet causes friction. Electrons move to the person and charge builds up. When the person touches a metal object, the electrons conduct away, making a spark.  
**Electric fields**  
Charged objects create electric fields around them. Strongest closest to the object. The field direction is the direction of force on a positive charge. Add more charge increases field strength.

Mechanical	Force acts upon an object
Electrical	Electric current flow
Heat	Temperature difference between objects
Radiation	Electromagnetic waves or sound

### Energy pathways

Kinetic energy	Energy stored by a moving object	$\frac{1}{2} \times \text{mass } X (\text{speed})^2$ $\frac{1}{2} mv^2$
Elastic Potential energy	Energy stored in a stretched spring, elastic band	$\frac{1}{2} \times \text{spring constant } X (\text{extension})^2$ $\frac{1}{2} ke^2$ <small>(Assuming the limit of proportionality has not been exceeded)</small>
Gravitational Potential energy	Energy gained by an object raised above the ground	Mass $X$ gravitational field strength $X$ height $mgh$

System	An object or group of objects that interact together	EG: Kettle boiling water.
Energy stores	Kinetic, chemical, internal (thermal), gravitational potential, elastic potential, magnetic, electrostatic, nuclear	Energy is gained or lost from the object or device.
Ways to transfer energy	Light, sound, electricity, thermal, kinetic	EG: electrical energy transfers chemical energy into thermal energy to heat water up.
Unit	Joules (J)	

Work	Doing work transfers energy from one store to another	By applying a force to move an object the energy store is changed.	Work done = Force $X$ distance moved $W = Fs$
Power	The rate of energy transfer	1 Joule of energy per second = 1 watt of power	Power = energy transfer $\div$ time $P = E \div t$ Power = work done $\div$ time, $P = W \div t$

Specific Heat Capacity	Joules per Kilogram degree Celsius ( $J/Kg \text{ } ^\circ C$ )
Temperature change	Degrees Celsius ( $^\circ C$ )
Work done	Joules (J)
Force	Newton (N)
Distance moved	Metre (m)
Power	Watts (W)
Time	Seconds (s)

Units		Useful energy	Wasted energy
Prefix	Multiple	Energy transferred and used	Dissipated energy, stored less usefully
Kilo	1000		
Mega	1000 000		
Giga	100 000 000		
		Standard form	
		$10^3$	$10^6$
		$10^9$	

Change in thermal energy = mass  $X$  specific heat capacity  $X$  temperature change

Specific Heat Capacity	Energy needed to raise 1kg of substance by 1 $^\circ C$	Depends on: mass of substance, what the substance is and energy put into the system.
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## Energy stores and changes

## AQA ENERGY - part 1

Closed system	No change in total energy in system
Open system	Energy can dissipate

## Energy Conservation and Dissipation

**HIGHER:** When an object is moved, energy is transferred by doing work.

Work done = Force  $X$  distance moved

Frictional forces cause energy to be transferred as thermal energy. This is wasted.

$$\Delta E = m \times c \times \Delta \theta$$

**HIGHER:** efficiency can be increased using machines.

$$\text{Efficiency} = \frac{\text{Useful power output}}{\text{Total power input}}$$

$$\text{Efficiency} = \frac{\text{Useful output energy transfer}}{\text{Total input energy transfer}}$$

Efficiency: How much energy is usefully transferred

**Dissipate**  
To scatter in all directions or to use wastefully  
When energy is 'wasted', it dissipates into the surroundings as internal (thermal) energy.



Ways to reduce 'wasted' energy: Energy transferred usefully  
Insulation, streamline design, lubrication of moving parts.

**Principle of conservation of energy**  
The amount of energy always stays the same.  
Energy cannot be created or destroyed, only changed from one store to another.

Energy (KE, EPE, GPE, thermal)	Joules (J)
Velocity	Metres per second (m/s)
Spring constant	Newton per metre (N/m)
Extension	Metres (m)
Mass	Kilogram (kg)
Gravitational field strength	Newton per kilogram (N/kg)
Height	Metres (m)

Reducing friction - using wheels, applying lubrication. Reducing air resistance - travelling slowly, streamlining.

Each Kg has a gravitational pull of 9.8N.

Gravitational field strength	Gravity exerted around an object.	Earth's $g_{fs} = 9.8N/kg$
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Unit	Newton (N)	1N
Kilo	Kilonewton (kN) = 1000	$1 \times 10^3$
Mega	Meganewton (MN) = 1000,000	$1 \times 10^6$

Centre of mass acts through a single point

Weight = mass X gravitational field strength  
 $W = m \times g$

Weight	Force acting upon an object due to gravity	Newton (N)
Mass	How much matter	Kilograms (kg)

**Scalar**  
A quantity that only has magnitude (size)  
e.g. mass, time, speed, temperature, energy.

**Vector**  
A quantity that only has magnitude and direction  
e.g. force, velocity, momentum

An arrow can be used to show vectors

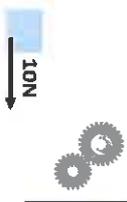
Length of arrow = magnitude of vector  
Direction of arrow = direction of vector

Velocity	Speed + direction	The speed of a car is 30m/s. A car moves forward with a velocity of 30m/s
Distance	How far	The table is 1m long
Displacement	Distance + direction	The beach is 1km due east of the town

Area	Metres squares ( $m^2$ )
Weight	Newton (N)
Mass	Kilograms (kg)
Gravitational field strength	Newton per kilogram (N/kg)
Force	Newton (N)
Work done	Joules (J)
Distance	Metres (m)
Moment	Newton-metres (Nm)

Pressure = height X density X  $g_{fs}$

**Gravity**



Moments, levers and gears

Moment = force X distance

**Moment**  
Turning effect of a force about a pivot

**Lever**  
A small force exerted with a long lever applies a large force

Gears  
Increase or decrease the rotational effect of a force

**HIGHER ONLY**  
Pressure

Pressure = Force ÷ Area  
 $P = F \div A$

Pressure and depth  
Pressure on divers depends on weight of water above

Uphrust  
Resultant force exerted by a fluid

Fluid  
A liquid or gas  
Flows and changes shape to fill a container.  
Use liquids to transmit pressure  
Hydraulic machine

Atmospheric pressure  
Caused by billions of air particles colliding with a surface.

**AQA FORCES - part 1**

**Forces and their interactions**  
Scalar and vector quantities

**Resultant force**  
The overall effect of all of the forces acting upon an object

Two forces acting in the same direction are added.  
Two forces acting in the opposite direction are taken away.

**Higher ONLY**  
Work done against frictional forces, temperature of object rises.

**Free body diagram**  
Show magnitude and direction of all forces upon an object

Object moves left with a force of 5N

Force	Push or pull	Stretch, squash, turn.
Contact force	Exerted between two objects when they touch	Friction, air resistance, tension.
Non-contact force	Exerted between two objects without touching	Gravity, electrostatic forces, magnetic forces.

**Resolving forces**  
An object pulled with a force at an angle  
A single force can be split into two components acting at right angles to each other.

The component forces combined have the same effect.

**Work done and energy transfer**

**Work done**  
When work is done, energy is transferred

1J of work is done when 1N of force moves an object through a distance of 1m, in the direction of the force.

Work done = force X distance moved  
 $W = F \times s$

**Forces and elasticity**

Elastic deformation	The object has been stretched but returns to its original length	One force More than one force	The object changes speed or direction The object changes shape	Two balanced forces can stretch an object. Two balanced forces can compress an object. Three balanced forces can bend an object.
Inelastic deformation	The object has been stretched but does not return to its original length			
Extension	The difference between stretched and unstretched lengths			Limit of proportionality Beyond this point the spring is permanently deformed



Stretching a spring  
Force = spring constant X extension,  $F = k \times e$   
 $EPE = \frac{1}{2} \times \text{spring constant} \times (\text{extension})^2$ ,  $EPE = \frac{1}{2} k e^2$

Elastic Potential energy (EPE)  
Energy stored in a stretched spring

Force	Newton (N)
Spring constant	Newton per metre (N/m)
Extension	Metres (m)
EPE	Joules (J)

**Relay**  
A device using a small current to control a larger current in another circuit

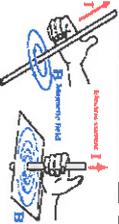
Solenoid is wound around an iron core. Small current magnetises the solenoid. This attracts to electrical contacts, making a complete circuit. Current flows from battery to starter motor.

**Electromagnet**  
Lots of turns of wire increase the magnetising effect when current flows  
Turn current off, magnetism lost.

Increase strength of magnetic field  
Use larger current  
Use more turns of wire  
Put turns of wire closer together  
Use iron core in middle

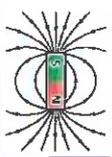
**Solenoid**  
A long coil of wire  
Magnetic field from each loop adds to the next.  
Reverse current, magnetic field direction reverses.  
Further away from the wire, magnetic field is weaker.  
Current large enough, iron filings show circular magnetic field.  
If current is small, magnetic field is very weak.

**Right hand rule**  
Thumb: Direction of current.  
Fingers: Direction of magnetic field.



Electric current flowing in a wire produces a magnetic field around it.

Magnetic field around a wire



**Permanent and Induced Magnetism**

Magnets	Magnets	Magnets
<b>Magnetic</b>	Materials attracted by magnets	Uses non-contact force to attract magnetic materials.
North seeking pole	End of magnet pointing north	Compass needle is a bar magnet and points north.
South seeking pole	End of magnet pointing south	Like poles (N - N) repel, unlike poles (N - S) attract.
Magnetic field	Region of force around magnet	Strong field, force big. Weak field, force small. Field is strongest at the poles.
Permanent	A magnet that produces its own magnetic field	Will repel or attract other magnets and magnetic materials.
Induced	A temporary magnet	Becomes magnet when placed in a magnetic field.

**Split-ring commutator**  
Slip ring touching two carbon brush contacts

**Loud speakers**  
Converts variations in electrical current into sound waves.

Varying current flows through a coil that is in a magnetic field. A force on the wire moves backwards and forwards as current varies. Coil connected to a diaphragm. Diaphragm movements produce sound waves.

**Generators**  
Coil of wire rotating inside a magnetic field. The end of the coil is connected to slip rings.  
Produces alternating current.

**Microphones**  
Converts pressure variations in sound waves into variations in current in electrical circuits.

**Electric motor**  
Coil of wire rotates about an axle  
Current flows through the wire causing a downward movement on one side and an upward movement on the other side.

**Fleming's left-hand rule**  
To predict the direction a straight conductor moves in a magnetic field.

Thumb	Direction of movement.
First finger	Direction of magnetic field.
Second finger	Direction of current.



$F = B \times I \times L$   
Force = magnetic flux density X current X length

If current and magnetic field are parallel to each other, no force on wire.

**Motor effect**  
HIGHER only  
AQA  
MAGNETISM AND ELECTROMAGNETISM

Induced potential, transformers and National Grid

Magnetic fields from the permanent magnet and current in the foil interact. This is called the motor effect.  
Reverse the current, foil moves upwards.  
Aluminium foil placed between two poles of a strong magnet, will move downwards when current flows through the foil.  
Size of force acting on foil depends on magnetic flux density between poles, size of current, length of foil between poles.

Magnetic flux density	Lines drawn to show magnetic field	Lots of lines = stronger magnets.
Magnetic flux	Number of lines of magnetic flux in a given area	Measures the strength of magnetic force.

National Grid  
Distributes electricity generated in power stations around UK

**PHYSICS HIGHER only**

**Transformer**  
Two coils of wire onto an iron core  
Alternating current supplied to primary coil, making magnetic field change. Iron core becomes magnetised, carries changing magnetic field to secondary coil. This induces p.d.

Step-up transformers  
Increase voltage, decrease current

Step-down transformers  
Decrease voltage, increase current

Increases efficiency by reducing amount of heat lost from wires.  
Makes safer value of voltage for houses and factories.

**Induced potential**  
When a conducting wire moves through a magnetic field, p.d. is produced

Power lost = Potential difference X Current

Power supplied to primary coil = power supplied to secondary coil  
 $V_p \times I_p = V_s \times I_s$

Voltage across the coil X number of coils (primary) = Voltage across the coil X number of coils (secondary)  
 $V_p \div V_s = n_p \div n_s$

**Generator effect**  
Generates electricity by inducing current or p.d.

Uses of the generator effect  
Dynamo, Microphones

Force	Newton (N)
Magnetic flux density	Tesla (T)
Current	Amperes (A)
Length	Metres (m)
Power	Watts (W)
p.d.	Voltage (V)



Pressure of a fixed volume of gas increases as temperature increases (temperature increases, speed increases, collisions occur more frequently and with more force so pressure increases).

Temperature of gas is linked to the average kinetic energy of the particles.

If kinetic energy increases so does the temperature of gas.

No kinetic energy is lost when gas particles collide with each other or the container.

Gas particles are in a constant state of random motion.

$$P = m \div V$$

Density = mass ÷ volume.

Density **Mass of a substance in a given volume**

Freezing  
Liquid turns to a solid. Internal energy decreases.

Melting  
Solid turns to a liquid. Internal energy increases.

Boiling / Evaporating  
Liquid turns to a gas. Internal energy increases.

Condensation  
Gas turns to a liquid. Internal energy decreases.

Sublimation  
Solid turns directly into a gas. Internal energy increases.

Conservation of mass  
When substances change state, mass is conserved.

Physical change  
No new substance is made, process can be reversed.

State	Particle arrangement	Properties
Solid	Packed in a regular structure. Strong forces hold in place so cannot move.	Difficult to change shape.
Liquid	Close together, forces keep contact but can move about.	Can change shape but difficult to compress.
Gas	Separated by large distances. Weak forces so constantly randomly moving.	Can expand to fill a space, easy to compress.

### Particle model

### AQA PARTICLE MODEL OF MATTER

#### Pressure

**PHYSICS ONLY:** when you do work the temperature increases e.g. pump air quickly into a ball, the air gets hot because as the piston in the pump moves the particles bounce off increasing kinetic energy, which causes a temperature rise.

$$PV = \text{constant.}$$

$$P_1 V_1 = P_2 V_2$$

Change in thermal energy = mass X specific heat capacity X temperature change.

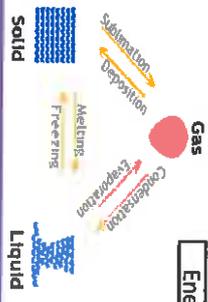
$$\Delta E = m \times c \times \Delta \theta$$

#### Internal energy and energy transfers

Specific Latent Heat	Energy needed to change 1kg of a substance's state
Specific Latent Heat of Fusion	Energy needed to change 1kg of solid into 1 kg of liquid at the same temperature
Specific Latent Heat of Vaporisation	Energy needed to change 1kg of liquid into 1 kg of gas at the same temperature

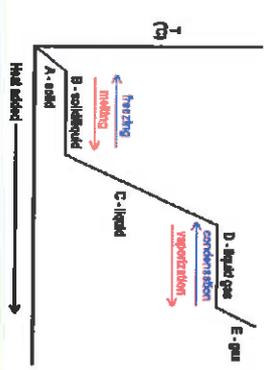
Energy needed = mass X specific latent heat.

$$\Delta E = m \times L$$



	Units
Density	Kilograms per metre cubed (kg/m³)
Mass	Kilograms (kg)
Volume	Metres cubed (m³)
Energy needed	Joules (J)
Specific latent heat	Joule per kilogram (J/kg)
Change in thermal energy	Joules (J)
Specific heat capacity	Joule per kilogram degrees Celsius (J/kg °C)
Temperature change	Degrees Celsius (°C)
Pressure	Pascals (Pa)

Internal energy	Energy stored inside a system by particles	Internal energy is the total kinetic and potential energy of all the particles (atoms and molecules) in a system.
Heating changes the energy stored within a system	Heating causes a change in state. As particles separate, potential energy stored increases. Heating increases the temperature of a system. Particles move faster so kinetic energy of particles increases.	





Milky Way our galaxy.

Planet	<i>A large body orbiting the Sun</i>
Moon	<i>A natural satellite orbiting a planet</i>
Dwarf planet	<i>A body large enough to have its own gravity which caused a spherical shape</i>
Solar system	<i>Any object orbiting the Sun due to gravity</i>
Galaxy	<i>Collection of billions of stars</i>
Universe	<i>Collection of galaxies</i>



Solar system  
Comets, asteroids, satellites  
Other objects.

**The life cycle of a star.**

Nebula	<i>A cloud of cold hydrogen gas and dust</i>	Cloud collapses due to gravity, particles move very fast colliding with each other, kinetic energy transfers into internal energy and the temperature increases.
Protostar	<i>The large ball of gas contracts to form a star</i>	High temperature causes Hydrogen nuclei to collide and nuclear fusion begins. A star is 'born'.
Main sequence	<i>Stable period of star</i>	Gravity tries to collapse the star but enormous pressure of fusion energy expands and balances the inward force.



**Stars the same size as our Sun.**

Red giant	<i>A large star that fuses Helium into heavier elements</i>	Hydrogen runs out, star becomes unstable, pressure inside drops causing star to collapse. Atoms now closer together results in atoms fusing and temperature increases. This increase in temperature causes the core to swell.
White dwarf	<i>Star collapses</i>	Nuclear fuel runs out, fusion stops, dense very hot core.
Black dwarf	<i>Cold dark star</i>	White dwarf cools down.

**Stars larger than our Sun.**

Red super giant	<i>Star swells greatly</i>	Nuclear fuel begins to run out and star swells (more matter = bigger size).
Supernova	<i>Gigantic explosion due to run away fusion reactions</i>	Rapid collapse, heats to very high temperatures causing run away nuclear reactions, star explodes, flinging remnants out into space. Large gravitational forces collapse the core into a tiny space. Remains of supernova form heavier elements (iron and above)
Neutron star	<i>Very dense star</i>	Made out of neutrons.

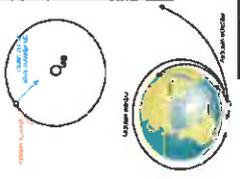
**Effect of gravity.**

Gravity causes moons to orbit planets, planets to orbit the Sun, stars to orbit galaxy centres.  
Force of gravity changes the moon's direction not its speed.  
Gravity pulls objects towards the ground.

**Orbital motions**

**Speed of Orbit.**

Too fast = disappears into space.  
Correct speed = steady orbit around Earth.  
Too slow = falls to Earth.  
To calculate speed of Orbit: distance object moves in 1 orbit, Distance =  $2\pi r$ , then average speed = distance ÷ time.



**HIGHER:**

Velocity = a vector.  
A planet's velocity changes but speed remains constant.

Due to the Sun's gravity, planets accelerate towards the Sun and so changes direction.

**HIGHER: Circular orbits.**

When ambulances go past the sound changes from a high pitch to a low pitch.

Planets close to the Sun, gravity pull is strong. Planets move quickly.  
Planets further away from the Sun, gravity pull is weaker. So speed of planet is slower.

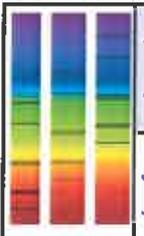
Frequency of sound wave decreases, wavelength increases.

**AQA SPACE PHYSICS PHYSICS ONLY**

**Red shift**

**Understanding models.**

Red-shift	<i>The observed increase in wavelength of light from most distant galaxies. Light moves towards the red end of the spectrum.</i>
Hubble (1929)	<i>He studied light from distant galaxies; found as frequency decreases, wavelength increases.</i>
The Big Bang	<i>Universe began 13.8 billion years ago</i>



All matter and space expanded violently from a single point.  
Red-shift provides evidence for expansion.

Aristotle (ancient Greek)	<i>Earth at the centre, other heavenly bodies move around the Earth.</i>
Copernicus (1473 - 1543)	<i>Sun at the centre, other heavenly bodies move around the Sun.</i>
Galileo (1610)	<i>Made a telescope, looked at Jupiter, found four moons rotating around planet.</i>

Galaxies are moving away from us in all directions.  
Light from distant galaxies is red-shifted, so galaxy is moving away from us.  
Galaxies further away have bigger red-shift so are moving faster away.

Planets and moons moved at different speeds to stars = reason for different positions.

**OR if collapse is into a really tiny space.**

Black hole

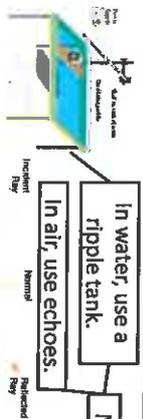
No light escapes

Gravitational forces so strong everything is pulled in.



Wave speed	Wave speed = frequency X wavelength	$V = f \times \lambda$
Wave period	Wave period = $1 \div$ frequency	$T = 1 \div f$
Speed	Speed = distance $\div$ time	$v = d \div t$

Wavelength	Distance from one point on a wave to the same point of the next wave
Amplitude	The maximum disturbance from its rest position
Frequency	Number of waves per second
Period	Time taken to produce 1 complete wave



In water, use a ripple tank.  
In air, use echoes.  
Measuring speed  
Sound waves travelling through different mediums, the frequency stay constant.

Angle of incidence = angle of reflection  
 $(i) = (r)$

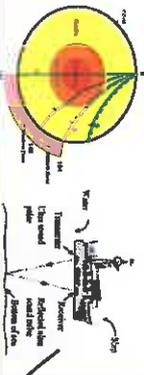
Reflection	Wave bounces off the surface.
Refraction	Waves changes direction at boundary.
Transmitted	Passes through the object.
Absorbed	Passes into but not out of, transfers energy and heats up the object.



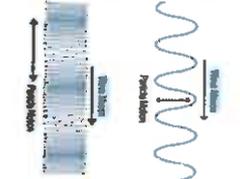
**PHYSICS HIGHER ONLY**  
Frequencies between 20 - 20,000 Hz  
Hearing

Longitudinal waves cause ear drum to vibrate, amplified by three ossicles which creates pressure in the cochlea.

P wave	S wave	Seismograph
Longitudinal	Transverse	Shows P and S waves arriving at different times.
Fast	Slow	
Travel through liquids and solids	Travels through solids	
Produced by earthquakes.		By using the times the waves arrive at the monitoring centres, the epicentre of earthquakes can be found. ( $v = x \div t$ ).



Ultra sound	Partially reflected off boundary	Used for medical and foetal scans.
Sonar	Reflected off objects	Used to determine depth of objects under the sea.



Transverse wave	Vibration causing the wave is at right angles to the direction of energy transfer	Energy is carried outwards by the wave.	Water and light waves, S waves.
Longitudinal wave	Vibration causing the wave is parallel to the direction of energy transfer	Energy is carried along the wave.	Sound waves, P waves.

Transverse and Longitudinal waves

Waves in air, fluids and solids

AQA Waves

Electromagnetic waves

Electromagnetic wave spectrum  
gamma ray, ultraviolet, visible, infrared, radio

Absorbed light changes into thermal energy store.

Black body radiation

Short wavelengths have high frequency and high energy.  
e.g. Gamma

PHYSICS ONLY  
Earth and Global warming

Black body radiation  
Constant temperature  
Rate of absorption = rate of radiation

Ultraviolet, visible light, infra-red radiation penetrate atmosphere and heat up Earth's surface.  
Longer wavelengths are radiated back, trapped by atmosphere.

Energy lost is not at the same rate as energy being absorbed so Earth heats up.

Hotter objects emit more infrared radiation.  
Intensity and wavelength of energy affects temperature.



Magnification = image size  $\div$  object size  
PHYSICS ONLY  
HIGHER: Lenses

HIGHER: Properties  
Concave  
Convex

Concave	Real or virtual images.	Only virtual images.	$2F$	Image same size, upside down, real.
Convex	Only virtual images.	$2F - F$	$< F$	Image larger, upside down, real. Image bigger, right way, virtual.

Diffuse	Specular
Rough surface reflection.	Flat surface reflection.

Black surfaces	Good absorbers, good emitters.
White surfaces	Poor absorbers, poor emitters.
Shiny surfaces	Good reflectors
EM waves refract	

EM wave	Danger	Use
Radio	Safe.	Communications, TV, radio.
Microwave	Burning if concentrated.	Mobile phones, cooking, satellites.
Infrared	Damage to eyes.	Heating, remote controls, cooking.
Visible	Sunburn, cancer.	Illumination, photography, fibre optics.
Ultra violet	Cell destruction, mutation, cancer.	Security marking, disinfecting water.
X-ray		Broken bones, airport security.
Gamma		Sterilising, detecting and killing cancer.

Low frequency, long wavelength.	White Wave lengths reflected
High frequency, short wavelength.	Black Wave lengths absorbed